Effect of Oil Contamination on the Electrical Resistivity and Rheological Properties of Smart Spacer Fluid

M. Sai Anudeep Reddy and C. Vipulanandan, Ph.D., P.E. Texas Hurricane Center for Innovative Technology (THC-IT) Department of Civil and Environmental Engineering University of Houston, Houston, Texas 77204-4003 E-mail:srmaddi@uh.edu, cvipulanandan@uh.edu Phone: (713) 743-4278

Abstract: In this study, the spacer fluids were tested for rheological properties with Oil contamination varying from 0 to 5%. The spacer rheology was modelled using Bingham-plastic model, Herchel Burlkey model and Vipulanandan model. The spacer fluid with 1% nanoFe₂O₃ showed increase in rheological properties with the increase in Oil contamination from 0 to 5%. The Plastics viscosity increased from 26 to 47cP, a 81% increase and yield point from 6 Pa to 12.7 Pa, a 112% increase. The electrical resistivity changed from 0.19 Ω -m to 7.8 Ω -m, a 400% increase with 5% Oil contamination.

1. Introduction:

Spacers or Flushes are used to thin and disperse drilling-fluid particles, and are used to separate drilling fluids and cementing slurries. These spacers can be used with either water based or oil based drilling fluids. Use of this spacer fluid prepares the casing and the formation for cementing process. Recent studies have shown that when nanomaterials are added to the drilling muds it can be used as a sensing material downhole for temperature and pressure[1]. Use of nanomaterials in spacer fluids enhances cleaning ability and can make it sensing. There is no current method to monitor the contamination of spacer fluid in real time. Electrical resistivity can be used as sensing parameter for monitoring the extent of contamination in the spacer fluids[2].

2. Objective: The main objective was to quantify the effect of oil contamination on electrical resistivity and rheological properties of smart spacer fluid.

3. Materials and Method:

The spacer fluid was prepared by using water as base fluid, Rheology Modifier as 0.75% Guargum, Surfactant: 0.4% UHBS, Inhibitor: 3% KCl, with a Modification: Nano Iron (Fe_2O_3) : 1%. KCl was mixed in base fluid water and mixed thoroughly until dissolved. Then rheology modifier Guargum was added followed by UH Bio-surfactant and mixed until uniform solution is obtained. Nano iron is then added as a modification.

Modeling:

Bingham plastic Model:

Two parameters model which can be presented as followed:

 $\tau = \tau_y + \eta_p \dot{\gamma}$, in which τ_y is yield stress and η_p is plastic viscosity of the fluid.

Herschel-Bulkley Model:

This is a three parameters model represented as followed:

 $\tau = \tau_{0_1} + k\dot{\gamma}^n$, Where τ_0 is yield stress and k and n are experimentally fit parameters. Vipulanandan Model:

$$\tau = {\tau_0}_2 + \frac{\dot{\gamma}}{A+B\dot{\gamma}}$$

in which τ_0 is yield stress and k and n are experimentally fit parameters. If we calculate the ultimate shear stress from this model we will have:

 $\lim_{\dot{\gamma}\to\infty}\tau=\tau_{0_2}+\frac{1}{B}$

4. Results and Discussion:

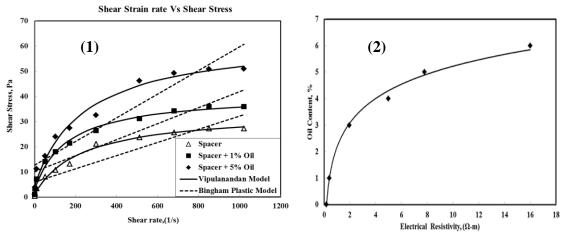


Figure 1- Effect of Oil Contamination on Rheology of spacer with NanoFe₂O₃ Figure 2- Effect of Oil Contamination on Electrical Resistivity

Table 1: Kneological Model parameters for effect of On Contamination											
	Bingham Plastic Model			Hershel Burkely Model			Vipulanandan Model				
Model Parameters	PV	PV(cP)	YP(Pa)	n	k	τ (yield)	A(Pa. s) ⁻¹	B (Pa) ⁻¹	τ (yield)(Pa)	τ (max)(Pa)	$\Delta \tau$ (max)(Pa)
Spacer + NanoFe	0.026	26	6	0.41	1.72	0	7.1	0.03	1.25	34.6	0
1% Oil	0.031	31	10.1	0.34	3.62	0	4.21	0.026	3.01	41.5	19.9
5% Oil	0.047	47	12.7	0.37	4.02	0	4.15	0.017	4.57	63.4	52.9

 Table 1:Rheological Model parameters for effect of Oil Contamination

Bingham model (1919)

The spacer fluid with 1% nanoFe₂O₃ showed increase in rheological properties with the increase in Oil contamination from 0 to 5%. The Plastics viscosity increased from 26 to 47 cP and yield point from 6 Pa to 12.7 Pa. (Fig. 1 and Table 1)

Vipulanandan model (2014)

Increasing the percentage of oil contamination increased the τ max values by 50%. For upto 5% contamination the τ max values were found to increase linearly. The yield stress also increased from 1.25 Pa to 4.57 Pa with 5% Oil contamination.

Resistivity: The electrical resistivity was found to increase with the increase in the oil content. The resistivity increased from 0.19 Ω -m to 7.8 Ω -m for 5% oil contamination. It tends to reach a maximum value which after which any further addition of oil does not change the resistivity. This can be termed as the maximum oil limit taken by the Spacer solution (Fig. 2).

5. Conclusion:

Electrical resistivity can be used to monitor the percentage of contamination in the spacer fluid which changes by 400% for 5% contamination. The maximum shear stress also increased from 34.6 pa to 63.4 Pa, a 53 % increase with 5% oil contamination.

6. Acknowledgements:

This study was supported by the Texas Hurricane Center for Innovative Technologies (THC-IT), University of Houston, Houston, Texas.

7. References:

- 1. Fakoyi, M. and Shah, S.," Emergence of Nanotechnology in the Oil and Gas Industry: Emphasis on the application of Silica Nanoparticles", Petroleum, 2017.
- 2. Mohammed, A.S., "Effect of Temperature on the Rheological Properties with Shear Stress Limit of Iron Oxide Nanoparticle Modified Bentonite Drilling Muds". Egyptian Journal of Petroleum, 2016.